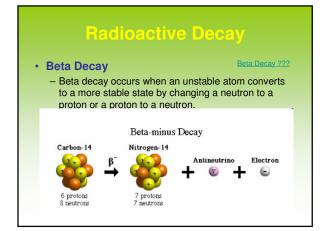


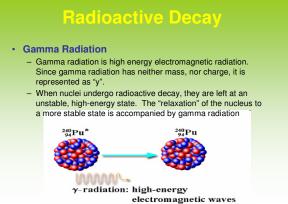
Radioactive Decay

- Alpha Decay
 - Alpha decay is a type of radioactive decay in which an atomic nucleus emits an <u>alpha particle</u> (two protons and two neutrons) and thereby transforms (or 'decays') into an atom with a mass number 4 less and atomic number 2 less.

 $^{226}_{88}Ra \rightarrow ^4_2He + ^{222}_{86}Rn$ Alpha Particle



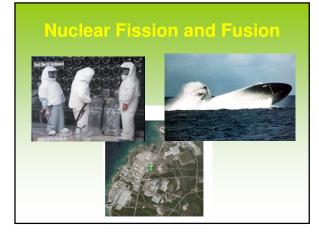




Radiation Poisoning

 Radioactive waste from nuclear reactions can lead to health problems. For example, ions of the radioactive isotope strontium-90, an alkali metal, exhibit chemical behavior similar to calcium ions. This leads to incorporation of ions into bone tissue, sending ionizing radiation into the bone marrow, possible causing leukemia.





Nuclear Fission and Fusion

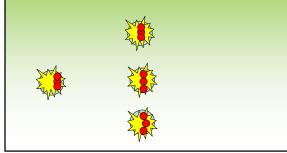
Nuclear Fission

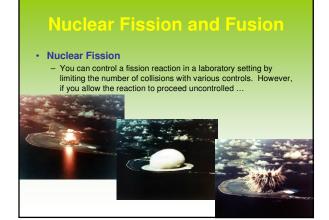
 Nuclear fission occurs when a highly unstable isotope splits into smaller particles. Nuclear fission usually has to be induced in a particle accelerator.

$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{87}_{35}Br + ^{146}_{57}La + 3^{1}_{0}n$$

Nuclear Fission and Fusion

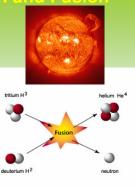
Nuclear Fission (A Chain Reaction)





Nuclear Fission and Fusion

- Nuclear Fusion
 - Nuclear fusion occurs when a target nucleus absorbs an accelerated particle. The reaction that takes place inside the Sun, or in a hydrogen bomb is a fusion reaction. Fusion reactions require extremely high temperatures but produce an enormous amount of energy.



Nuclear Fission and Fusion

- Nuclear Fusion
 - Unfortunately, to date, we have not yet been able to harness the energy in a fusion reaction to produce a huge amount of clean energy. However, there is hope to accomplishing this goal in the near future ...



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