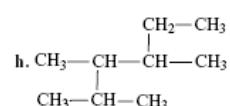
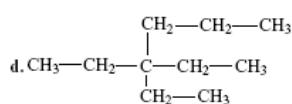
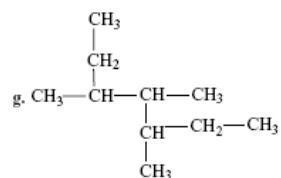
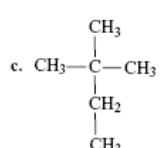
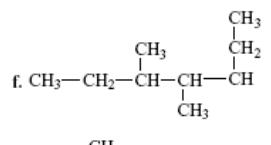
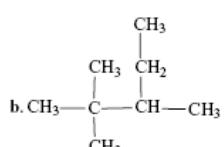
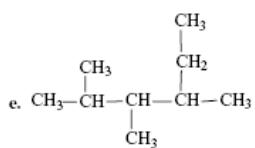
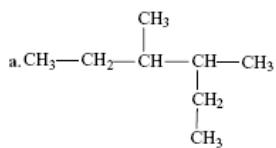
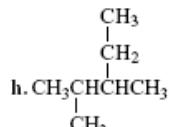
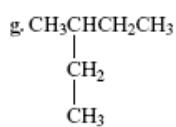
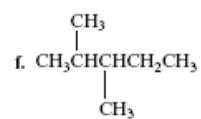
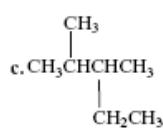
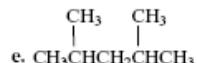
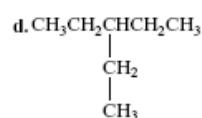
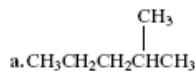


# HYDROCARBONS - REVIEW

1. Name the following hydrocarbons.



2. Name the following hydrocarbons.



5. Draw the structural formulas for the following:

a. 3-heptyne

b. cyclopentene

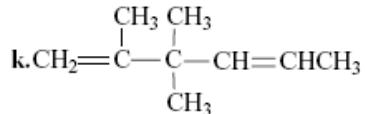
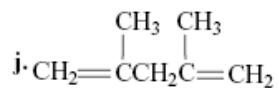
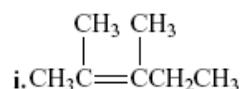
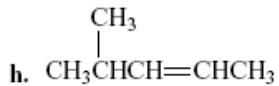
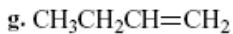
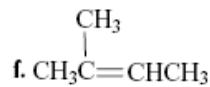
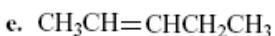
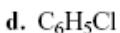
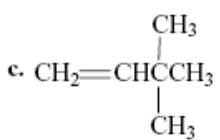
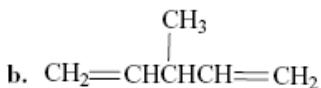
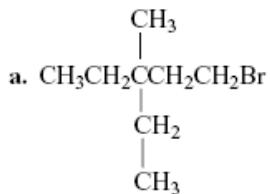
c. 3-phenyl-2,2-dimethylhexane

d. 1,3-butadiene

e. 1-ethyl-2-methylbenzene

f. 2,4-dimethyl-2-pentene

7. Name the following compounds.



8. Draw structural formulas for the following.

a. 3-heptene

e. 1,3-cyclopentadiene

c. trichloromethane

h. 2-bromo-3-methyl-2-butene

9. Write structural formulas for the following compounds.

a. 2-chlorobutane

i. 1-butanol

b. 2-butene

j. 3-methyl-2-pentene

c. 2-ethyl-3-methyl-1-butanol

k. 2-ethyl-4-methylpentanal

d. 3,3-dimethylbutanoic acid

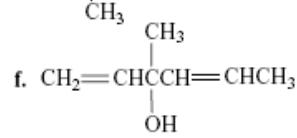
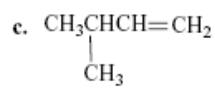
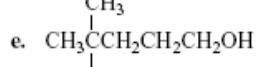
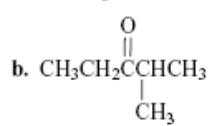
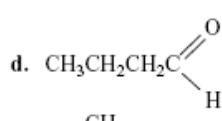
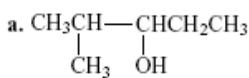
l. 3-ethyl-2,4-dimethyl-3-hexanol

e. 2,5,5-trimethyl-4-heptone

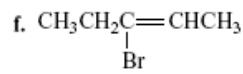
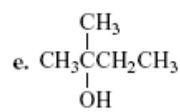
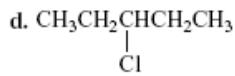
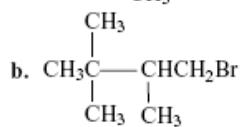
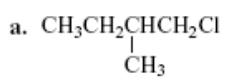
m. 5-chloro-3-ethyl-2-methylheptanoic acid

f. 1,8-nonadiyne

10. Name the following organic compounds.



11. Name the following organic compounds.



13. Draw structural formulas for the following.

a. Ethanal

f. propane

b. 2-butanone

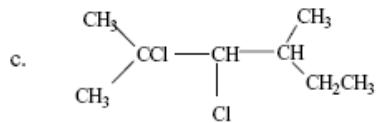
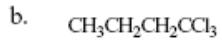
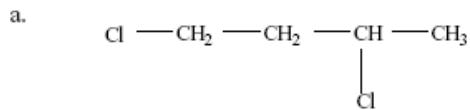
g. 2-pentyne

c. 2-methyl-2-propanol

h. cyclobutane

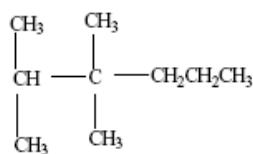
d. ethanoic acid

20. Name each of the following:

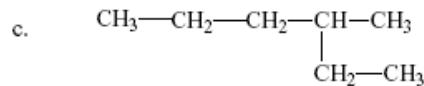
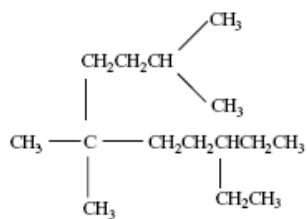


16. Name each of the following:

a.

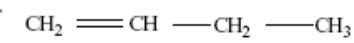


b.

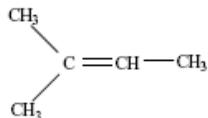


17. Name each of the following alkenes.

a.



b.



c.

